

**DOCUMENTS RELATING TO SESSIONAL
EXAMINARIONS, UNIT TESTS, HOME
ASSIGNMENTS, CLASS SEMINAR, Etc.**



**Principal I/c & Secretary
Barbhag College**

Department of Chemistry

Sessional Examination- 2022

B.Sc. 1st Semester

Sub : Chemistry(Hon) Paper: CHE-HE-1026

Full marks : 30

Time : 1Hour

1. Draw the Maxwell distribution of molecular velocity curves at different temperatures.
Also write the important feature of this curves 2+2=4
2. Calculate the vibrational degree of freedom of H_2O and CO_2 . 2
3. Find the expression of average velocity with the help of Maxwell distribution law . 4
4. Calculate the rms speed of N_2 gas at $270^\circ C$ and 70 cm pressure. (Given density of $Hg = 13.6 \text{ g/cm}^3$) . 3
5. Calculate the Miller Indices of crystal phases which cut through the crystal axis at
i) $(2a, 3b, c)$ ii) $(2a, -3b, -3c)$ 2
- 6 Calculate the number of atoms in per unit cell of fcc lattice. 2
- 7 Write the short notes on Radius Ratio. 3
- 8 The dissociation constant of Formic acid and Acetic acid are respectively 1.77×10^{-4} and 1.75×10^{-5} . Calculate the relative strength of two acids. 3
9. Calculate the p^H of solution obtained by mixing of $50 \text{ ml } 0.2 \text{ M HCl}$ with $50 \text{ ml } 0.1 \text{ M NaOH}$. 3
10. Calculate K_H for a salt of weak acid and strong base . Also establish a relationship among K_H , K_a and K_w for the same salt. 2+2=4

Principal /c & Secretary
Barbhag College

Unit Test(I)
B.Sc. 1st Semester
Sub: Chemistry (HG/RC)

Paper CHE-HG/RC-1016

Time : 1 hour

Full Marks: 20

Name of Students... Dhara Raj Jyoti Kalita Roll No... 45

- হাইড্রজেন বর্ণালীৰ ব্যাখ্যা কোনটো তত্বই আগবঢ়াব পাৰে?
 - ভেৰ্ণ্টনৰ পৰমানু তত্ব
 - ব্যাডাৰফৰ্ডৰ আৰ্হি
 - ব'ৰৰ তত্ব
 - ডাৰ্ভগলিৰ প্ৰকল্প
- হাইড্রজেন বর্ণালীৰ ধৰ্মা শ্ৰেণীৰ সংখ্যা
 - 4 টা
 - 5 টা
 - 2 টা
 - 6 টা
- বামৰ শ্ৰেণীক পোৱা হাইড্রজেন বর্ণালীৰ অংশ হ'ল-
 - অবলোহিত (IR)
 - অতিবেঙুনীয়া (UV)
 - দৃশ্যমান (Visible)
 - বর্ণালীৰ বাহিৰৰ অংশ
- বিভবাৰ্গৰ ধ্ৰুবকৰ মান -
 - $1.3561 \times 10^6 \text{ m}^{-1}$
 - $2.5677 \times 10^5 \text{ m}^{-1}$
 - $4.2121 \times 10^3 \text{ m}^{-1}$
 - $1.0967 \times 10^7 \text{ m}^{-1}$
- ব'ৰৰ তত্বটো প্ৰযোজ্য
 - 2 টা ইলেকট্ৰনযুক্ত তন্ত্ৰৰ বাবে
 - 1 টা ইলেকট্ৰনযুক্ত তন্ত্ৰৰ বাবে
 - 4 টা
 - 3 টা
- ব'ৰৰ আৰ্হিৰ বাবে 1 নং কক্ষপথত থকা ইলেকট্ৰনৰ বাবে হাইড্রজেন পৰমানুৰ ব্যাসাৰ্হ
 - 53 pm
 - 5.3 cm
 - 0.29m
 - 6.9 pm
- চুম্বকীয় (Magnetic) ক্ষেত্ৰত হাইড্রজেন বর্ণালীৰ ৰেখা বিভক্ত হোৱা পৰিঘটনাৰ নাম-
 - ষ্টাৰ্ক পৰিঘটনা
 - আলোক বিদ্যুৎ পৰিঘটনা
 - জিমান পৰিঘটনা
 - বিদ্যুৎ চুম্বকীয় প্ৰবেশ
- গতিশীল কৰা এটাৰ সৈতে জড়িত তৰংগ দৈৰ্ঘ্যক
 - দ্বৈত তৰংগ দৈৰ্ঘ্য
 - দ্য ভ্ৰগলিৰ তৰংগ দৈৰ্ঘ্য
 - প্ৰাংকৰ তৰংগ দৈৰ্ঘ্য
 - হাইড্রজেনবাৰ্গৰ তৰংগ দৈৰ্ঘ্য
- বৰ্ণ লেণ্ডে সমীকৰণ ব্যৱহাৰ কৰা হয়-
 - সহযোজী যোগৰ বিয়োজন শক্তিৰ বাবে
 - আয়নীকৰণ শক্তি নিৰ্ণয়ৰ বাবে (UV)
 - আয়নীকৰণ শক্তি নিৰ্ণয়ৰ বাবে
 - ইলেকট্ৰন বিভৱ নিৰ্ণয়ৰ বাবে
- কোনটোৰ গলনাংক আটাইতকৈ বেছি?
 - NaCl
 - NaF
 - NaI
 - NaBr
- SI এককত ID (ডিবাইব) মান
 - $3.335 \times 10^{-30} \text{ cm}$
 - $2.165 \times 10^{-25} \text{ Cm}$
 - $4.216 \times 10^{-21} \text{ Cm}$
 - $1.027 \times 10^{-19} \text{ Cm}$
- কোনটো অনু ধ্ৰুৱীয় নহয়?
 - NH₃
 - H₂O
 - HCl
 - CH₄

P.T.O.

- আয়নীকৰণ যৌগৰ সহযোজী ধৰ্ম নিৰূপণৰ কোনটো বিধি বা নিয়ম প্ৰয়োগ কৰা হয়?
 - শুণ্ৰ নিয়ম
 - বৰ্ণ হেৰাবৰ নীতি
 - হকেলৰ নীতি
 - ফাজানৰ নিয়ম
- মিথেন (CH₄) ত H-C-H বাহানি কোনৰ পৰিমাণ-
 - 120°
 - 107°
 - 104°
 - 109°28'
- এলকেনে কোনটো সমযোজিতা দেখুৱায়?
 - অৱস্থান সমযোজিতা
 - শুণ্ৰ সমযোজিতা
 - জ্যামিতীয় সমযোজিতা
 - আলোকীয় সমযোজিতা
- C₂H₄Br₂ আনৱিক সংকেত বিশিষ্ট গঠন সমযোজীৰ সংখ্যা-
 - 1
 - 2
 - 3
 - 4
- কাৰ্বক্সিলিক এচিডৰ ছিডিয়াম লৱন (RCOONa) সৈতে চডালাইম (NaOH + CaO) উত্তপ্ত কৰিলে কি উৎপন্ন হয়?
 - এলকেন
 - এলকহ'ল
 - এলকিন
 - এলডিহাইড
- 3-মিথাইল বিউট-1 ইনৰ গঠন-
 - CH₂ = CH-CH-CH₃
 - CH₃ - CH = CH - CH₃
 - CH₂ = CH-CH₂-CH₃
 - CH₂ = CH-CH₂-CH₃
- এলকেনৰ হেলোজেনেশ্বন (Halogenation) বিক্ৰিয়াত হেল'জেনবোৰৰ সক্ৰিয়তাৰ ক্ৰম-
 - Cl₂ > F₂ > I₂ > Br₂
 - F₂ > Cl₂ > Br₂ > I₂
 - I₂ > Br₂ > Cl₂ > F₂
 - Br₂ > I₂ > F₂ > Cl₂
- তলৰ কোনটোক উৰ্জৰ বিক্ৰিয়া বুলি কোৱা হয়?
 - CH₃CH₂I + HI $\xrightarrow{\text{ৰঙা P}}$ CH₃CH₃ + I₂
 - CH₃Br + 2Li $\xrightarrow{\text{শুকান ইথাৰ}}$ CH₃Li + LiBr
 - 2RX + 2Na $\xrightarrow{\text{শুকান ইথাৰ}}$ R-R + 2 NaX
 - CH₄ + Cl₂ $\xrightarrow{\text{পোহৰ}}$ CH₃Cl + HCl

Bharathi

Principal I/c & Secretary
Barbhag College

56

HOME ASSIGNMENT

Topic - Entropy

Name: Chandrama Talukdar

Sub: Chemistry Assignment

Class: B.Sc. 2nd semester

Class Roll NO: 38

Exam Roll NO: US-211-191-0009

Session - 2021-22

On 23/4/22

Principal I/c & Secretary
Barbhag College

Home Assignment

Topic - Entropy

06

B.Sc 2nd Semester (CBCS)

Name: Nabadeep Lahkar.

Sub: Physical Chemistry

Class Roll no: 34

Exam Roll no: US-211-191-0021

Reg. No: 21023050

Session - 2021-22

pd 23/4/22

Rishinchi

Principal /c & Secretary
Barbhag College

Department of Physics

Unit Test, 2021

Class: BSc. 1st Semester

Sub: Physics

Paper: PHY-HG/RC-1016

(Mechanics)

Total Marks: 15

Time: 40 Mins

1. Answer the following questions:

1x3=3

- Define dot product of two vectors.
- What do you mean by gravitational potential?
- What is the dimension of Poisson's Ratio?

2) Answer the following questions:

2x2=4

- A wire 0.5 m long and 1 sq mm in cross section has Young's modulus $1.24 \times 10^{11} \text{ N m}^{-2}$. How much work is done in stretching it through 1mm?
- A force $\vec{F} = -\vec{i} + 3\vec{j} + 4\vec{k}$ is acting at a point $5\vec{i} + 4\vec{j} + 3\vec{k}$. Obtain the moment of the force about the origin.

3) State the law of Gravitational attraction and hence define the Gravitational constant.

3

4) If Y, K and σ represent Young's Modulus, Bulk modulus and Poisson's Ratio respectively, Then prove that $K = \frac{Y}{3(1-2\sigma)}$.

5

Class: BSc. 4th Semester

Sub: Physics

Paper: PHY-HG/RC-4016

(Waves & Optics)

Total Marks: 15

Time: 40 Mins

1. Answer the following questions:

1x3=3

- Define spring constant.
- In Young's double slit experiment if D gets doubled what happens to the fringe width?
- State the necessary condition to observe the phenomenon of diffraction?

2) Answer the following questions:

2x2=4

- Write down two characteristics of S.H.M.
- Explain the nature of Newton's rings observed with white light.

3) Determine the expression for linear width of central maxima in the diffraction pattern due to a single slit.

3

4) What is Stoke's Law for the phase change of reflection? Show that a phase change of π occurs when the reflection takes place at the surface of a denser medium.

5

Principal i/c & Secretary
Barbhag College

Sessional Examination '21

B.Sc. 1st semester '21

PHYSICS
PHY-HG/RC-1016

Marks: 30

Time: 1 hour

The figures in the margin indicate full marks for the questions;

1. उत्तर दीजिए प्रत्येक प्रश्न को संक्षेप में।
Answer the following questions very shortly; 1×5=(5)

- (i) Define cross ~~(the)~~ product of two vectors.
- (ii) Give the example of 2nd order & 2nd degree of a differential equation.
- (iii) Write down the physical significance of cross product of two vectors.
- (iv) What do you mean by simple harmonic motion?
- (v) State the Kepler's third law.

2. Answer the following questions

2×5=(10)

- (a) Write down the postulates of special relativity theory.
- (b) Construct the differential equation of a SHM.
- (c) Find the general solution of 1st order homogeneous differential equation.
- (d) If $\vec{A} = 5\hat{i} - 7\hat{j} + 3\hat{k}$ and $\vec{B} = 6\hat{i} + 4\hat{j} - 2\hat{k}$ what is $\vec{A} \times \vec{B}$ and $\vec{A} \cdot \vec{B}$?
- (e) $y = \frac{dy}{dx} + \sqrt{1 + \left(\frac{dy}{dx}\right)^2}$, find out the order and degree from above equation.

T=0

Rinchi

Sessional Examination '21

B.Sc. 3rd sem

PHYSICS

Phy - HG/RC - 3016

Time: 60 min.

Marks: 30

The figures in the margin indicate full marks for the questions.

1x5=5

1. Give the short answers.

- state the 1st law of thermodynamics -
- Define only Carnot's cycle and Carnot's theorem.
- Write down the four thermodynamical potentials.
- State the Newton's law of cooling.
- Give the numerical value of solar constant.

2. Answer the following questions, 5x5=25

- State the relation $G = H + T \left(\frac{\partial G}{\partial T} \right)_P$ is a relation of Entropy and Gibbs potential.
- State and explain the mean free path on the basis of Einstein-Maxwell's concept.
- Give the relation between Entropy and Probability additive nature.
- State the Maxwell's thermodynamical relations.
- Explain the Maxwell's deduction of the value of the coefficient of viscosity.

Rishabh

Sessional Examination '22
B.Sc. 6th semester.

Marks 30 Physics
PHY-RE-6056 Time: 60 min.

The figures in the margin indicate full marks for the questions.

1. Answer the following questions shortly 2 × 5 = 10
- (a) What do you mean by length contraction
 - (b) Give the expression of Time dilation,
 - (c) State and explain the relativistic mass,
 - (d) State the postulates of special theory of Relativity
 - (e) What do you mean by Relativity of simultaneity
2. Answer the following questions. 4 × 5 = 20
- (i) Give the expression of acceleration of a particle of mass 'm' in uniform Electric field.
 - (ii) State and Explain the principle of virtual work.
 - (iii) Distinguish between the stable and unstable equilibrium system of an oscillating particle
 - (iv) Explain the relativistic term Minkowski space

B. Singh

Home Assignment
Physics

A

Name - Nabajyoti Deka
Roll no - US-211-191-0022
Regn. no - 21068007
class - BSc. 1st semester
Year - 2021-22

2

Biswas

B^+

Home Assignment

Name : Koushik Deba

Roll No. : US-211-191-0018

Reg. No. : 21068004

Class : BSc. 2nd Sem.

Subject : Physics (HGr)

Year : 2021-22

Principal I/c & Secretary
Barbhag College

A

Home Assignment

Name - Jubin Kumar Sarma

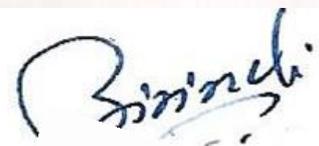
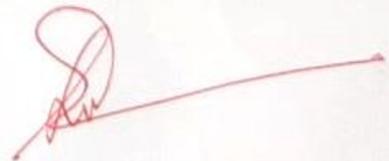
Roll no - US - 211 - 191 - 0015

Regn. no - 21068001

class - BSc. 2nd Semester.

Sub! - Physics

Session - 2021-22



Principal I/c & Secretary
Barbhag College

Department of Political Science

Sessional Examination-2022
Sixth Semester (Honours)
Political Science
Paper-POLHE 6016 (Human Rights)

Time: 1 hour

Full marks: 20

১) যিকোনো এটা প্ৰশ্নৰ উত্তৰ লিখা। ১× ১০

- ক) প্ৰাচীন ভাৰতত ধৰ্মৰ মাজেদি মানৱ অধিকাৰৰ ধাৰণা কেনেদৰে প্ৰকাশ পাইছিল আলোচনা কৰা।
খ) ৰাষ্ট্ৰীয় মানৱ অধিকাৰ আয়োগৰ গঠন, ক্ষমতা আৰু কাৰ্য আলোচনা কৰা।
গ) সশস্ত্ৰ বাহিনী বিশেষ ক্ষমতা (AFSPA) আইনখনৰ ইতিহাস আৰু ইয়াৰ মূল ব্যৱস্থা সমূহ বিশ্লেষণ কৰা।

৪) চমুটোকা লিখা (যিকোনো এটা) ১× ৫=৫

- (ক) কমনৱেলথ অৱ ইণ্ডিয়া বিল, ১৯২৫
(ক) ৰাষ্ট্ৰীয় মহিলা আয়োগ
(গ) নৰ্মদা বচাও আন্দোলন
(ঘ) ৰাজ্যিক মানৱ অধিকাৰ আয়োগ

৫) যিকোনো দুটা প্ৰশ্নৰ উত্তৰ লিখা ২× ২=৪

- ক) মানৱ অধিকাৰ আইনৰ যিকোনো দুটা ব্যৱস্থা উল্লেখ কৰা।
খ) ভাৰতীয় সংবিধানৰ মানৱ অধিকাৰ বুলিলে কি বুজা?
গ) উত্তৰ পূব ভাৰতত সন্ত্ৰাসবাদী কাৰ্যকলাপ উদ্ভৱৰ দুটা কাৰণ উল্লেখ কৰা।
ঘ) চিপকো আন্দোলনৰ সৈতে জড়িত দুগৰাকী উল্লেখযোগ্য ব্যক্তিৰ নাম লিখা। ২

৭) ৰাষ্ট্ৰীয় সংখ্যালঘু আয়োগ কিমান চনত গঠন হৈছিল? ১

- (খ) লোক প্ৰশাসনৰ প্ৰকৃতি
(গ) ওৱেবাৰৰ আমোলাতন্ত্ৰৰ বৈশিষ্ট্য

৪) লোক প্ৰশাসনৰ পৰিসৰ আলোচনা কৰা। 10

অথবা

বৈজ্ঞানিক পৰিচালনাৰ ধাৰণাটো আলোচনা কৰা। 10

অথবা

হেনৰী ফেয়লৰ প্ৰশাসনৰ ধাৰণাটো আলোচনা কৰা। 10

x x x x x x

Principal I/c & Secretary
Barbhag College

HOME ASSIGNMENT

TOPICS: লোক প্ৰশাসনৰ প্ৰকৃতি
আৰু পৰিষ্কাৰ আন্দোলন বন্ধা। (১)

Submitted by: Sanjay Das

ROLL NO : 133

subject : Political science (P.P.C)

class : B.A 5th semester

Gr.U. ROLL NO : UA-191-191-0108

Registration No: 19067990

Paper — POL RE 5016

OS

Rimochi

Principal I/c & Secretary
Barbhag College

BARBHAG COLLEGE

HOME ASSIGNMENT

05 RC

স্বল্প - নানাবিধ সামাজিক বিকাশ এবং পূর্বান কার্য

বিষয় - রাজনীতি বিজ্ঞান

নাম - শ্রী জয়ন্ত বেজ বসুয়া

শ্রেণী - স্নাতক প্রথম বর্ষ

শোলনং - ১০

বছর - ২০২২

Paper - POL - RE - 1016



Principal I/c & Secretary
Barbhag College

Schedule of Sessional Examination (Session: 2021-22)

B.Sc. 2nd and 4th semester {Honors and HG/RC (General)}

Department of Chemistry : Barbhag College

Date	Semester	Paper	Time	Room
14/07/2022(Thurs)	4 th semester (Honors)	CHE-HC-4016	10-11 AM	Honors Room
14/07/2022(Thurs)	2 nd semester (Honors)	CHE-HC-2016	12-1 PM	Room No 4
14/07/2022(Thurs)	2 nd semester HG/RC (General)	CHE-HG/RC-2016	12-1 PM	Room No 4
14/07/2022(Thurs)	4 th semester HG/RC (General)	CHE-HG/RC-4016	1-2 PM	Room No 4
15/07/2022(Fri)	4 th semester (Honors)	CHE-HC-4026	9-10 AM	Honors Room
15/07/2022(Fri)	2 nd semester (Honors)	CHE-HC-2026	10-11 PM	Honors Room
16/07/2022(Sat)	4 th semester (Honors)	CHE-HC-4036	10-11 AM	Honors Room



**HOD of Chemistry
Barbhag College
P.O.Katag, Nalanda, Amarnagar**

(Gautam Baishya)

HOD, Dept. of Chemistry

Barbhag College

07/07/2022



**Principal I/c & Secretary
Barbhag College**

SCHEDULE OF UNIT TEST (SESSION: 2021-22)

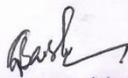
B.Sc. 1st / 3rd / 5th semester (H)

Date	Semester	Paper	Time
03/11/2021	1 st semester/ 3 rd semester	CHE-HC-1016/ CHE-HC-3016	9 AM
03/11/2021	5 th semester	CHE-HC-5016	10 AM
04/11/2021	1 st semester/ 3 rd semester	CHE-HC-1026/ CHE-HC-3026	9 AM
04/11/2021	5 th semester	CHE-HC-5026	10 AM
05/11/2021	3 rd semester/ 5 th semester	CHE-HC-3036/ CHE-HE-5026	9 AM
05/11/2021	5 th semester	CHE-HE-5056	10 AM

SCHEDULE OF UNIT TEST (SESSION: 2021-22)

B.Sc. 2nd / 4th / 6th semester (H)

Date	Semester	Paper	Time
02/03/2021	2 nd semester/ 4 th semester	CHE-HC-2016/ CHE-HC-4016	9 AM
02/03/2021	6 th semester	CHE-HC-6016	10 AM
03/03/2021	2 nd semester/ 4 th semester	CHE-HC-2026/ CHE-HC-4026	9 AM
03/03/2021	6 th semester	CHE-HC-6026	10 AM
04/03/2021	4 th semester/ 6 th semester	CHE-HC-4036/ CHE-HE-6026/	9 AM



Head of Chemistry
Barbhag College
P.O. Kalig, Nalbari (Assam)

Department of Chemistry
Barbhag College



Principal I/c & Secretary
Barbhag College

REPORT OF FIELD SURVEY
DEPARTMENT OF ECONOMICS
MARCH, 2022

The students (five) of B.A. 3rd semester with Economics honours conducted a survey in March, 2022 on the Socio-Economic Condition in different aspects such as livelihood sources, saving behaviour, Consumption pattern etc of rural people of the nearby villages of the college in the supervision of Dr. Nandita Goswami and Gitanjali Goswami. The students found that the average monthly income of the rural families is Rs. 34800 which varies from Rs. 16500 to Rs. 9000 with coefficient of variation of 101.6 percent, but the average monthly expenditure of the families is Rs. 17945 and it varies from Rs. 6250 to Rs. 37350 with coefficient of variation of 46.8 percent. Thus there is wide variation in income among the rural families but the difference is relatively less for expenditure. The rural families spend a major portion of their expenditure for food followed by travel, clothing, fuel (LPG and electricity), TV and mobile recharge, education, health etc. On an average, the rural families save almost 31 percent of their income and they mostly save their deposits in banks (85 percent) and post office (15 percent). The occupational structure of the rural families is shown in the following table.

Sl. no	Occupation	Percentage of families	
		As primary source	As secondary source
1	Farming	5	15
2	Government job	30	5
3	Private job	35	40
4	Business	10	5
5	Daily wage earning	20	35
6	Total	100	100

Source: Field survey

The table shows that the primary source of livelihood of 35 percent of the rural families is private sector job followed by Government job of 30 percent families. The secondary source of livelihood of 40 percent families is also private job followed by daily wage of 35 percent families. Moreover, one student of the said semester conducted her survey on the Self- Help groups of two villages of Barbhag area. She found five SHGs in villages Moura and Sonkuriha and each of them have ten members. Of these SHGs, three of are promoted by bank, one is by block and the other is self promoted. Three (60 percent) of these SHGs are engaged in economic activities like fish keeping, weaving and agricultural farming and the other two are engaged in accumulating savings of the members and in providing loans. The average annual profit of the SHGs engaging in economic activities is Rs. 29300.

The photos of the survey are given below:



Biswajit

Principal I/c & Secretary
Barbhag College

Rosworn
Head
Department of Economics
Barbhag College

CLASS SEMINAR: 2021-22
DEPARTMENT OF ECONOMICS, BARBHAG COLLEGE



MAIN THEME: ECONOMIC REFORM AND INDIAN ECONOMY

Date -22/9/2021

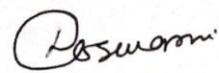
Report:

Today on 22/09/2021, Department of economics, Barbhag college has organized a class seminar among the students of B. A. Honours course. The main theme of the seminar is 'Economic Reform And Indian Economy'. Dr. Nandita Goswami, Head of the Department has chaired the seminar. Kanak Ch. Lahkar, former Head of the Department of Economics, Barbhag College was present in the seminar as resource person. Prof. Lahkar in his speech has briefly elaborated the rational and consequences of economic reform of India. He also has explained different problems facing by the Indian economy after economic reform of 1991. All about 10 students of the department has participated in the seminar. The two students of B. A. 3rd semester and one of B. A. 5th semester prepared their papers on three topics and presented in the seminar.

Jumi Kalita of B. A. 5th semester has presented her paper entitled "Globalization and Indian agriculture". In her paper she showed both the positive and negative impact of Globalization on Indian agriculture and discussed the opportunities and challenges which have been arose for globalization. The second paper presenter was Chandana Kalita of B. A. 3rd semester and title of the paper is 'Rational of economic reform in India'. In her paper Chandana discussed the factors and situation leading to the emergence of economic reform in India. The third paper presenter was Bhaskar Kalita of B. A. 3rd semester and the title of his paper is "Economic reform and IT sector of India". In the paper, he discussed the trend and development of IT sector of India after economic reform. Further he discussed both the challenges and opportunities arose to the sector after economic reform. At last the chair person Dr. Nandita Goswami expressed her opinion regarding the changing structure of the Indian economy and the problems faced by the economy after economic reform.




Principal i/c & Secretary
BARBHAG COLLEGE


Head
Department of Economics
Barbhag College

REPORT ON ESSAY COMPETITION
ORGANISED BY
DEPARTMENT OF CHEMISTRY, BARBHAG COLLEGE

Date: 28/05/2022

Today on 28/05/2022 an Essay Competition is held among the students of science stream, Barbhag College. The Competition is organized by the Department of Chemistry. The topic of the Essay Competition was "Ozone Layer Depletion and Green House Effect". 31 nos of students from different classes of science stream participated in the competition. All the faculty members of chemistry department forwarded necessary cooperation and help in successful organizing of the competition. Out of these participants three nos of students were selected as the 1st, 2nd and 3rd prize winners. The name of the prize winner students are given below-

1st prize --- Miss Sumita Rajbongshi (B.Sc. 4th semester Honors)

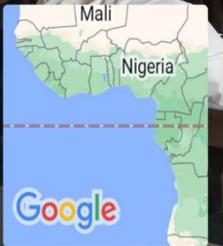
2nd prize--- Miss Chandrasmita Sarma (B.Sc. 2nd semester Honors)

3rd prize--- Miss Chandrama Talukdar (B.Sc. 2nd semester Honors)

Prizes of the competition were distributed on 14/06/2022.

Photographs and signature sheet are attached herewith





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Aparna
Principal i/c & Secretary
BARBHAG COLLEGE

Essay Competition 28/05/2022
 Department of Chemistry, Barbhag College
 Topic: Ozone Layer Depletion and Green House Effect

Name & Signature of Participating students:

1. Chandrasmita Sarma
2. Farja Rahman
3. Nabadeep Lahkar
4. Bhaskar Jyoti Das
5. Rupjyoti Barman
6. Bebek Mardindar
7. Hemangshu Patra
8. Nabadeep Kalita
9. Sumita Roshengshi
10. Manash Jyoti Baisya
11. Chandrama Talukdar
12. ~~Sumita~~
13. Jubin Kumar Sarma
14. Jyotismita Talukdar
15. Bipakshi Talukdar
16. Nidashree Kalita
17. Firdousi Rahman
18. Suman Baisya
19. Saucav Saikia
20. Moumoy Bisbaruah
21. Nisha Jyoti Kalita
22. Kaushik Saikia
23. ~~Sumita~~ Maja
24. Maniraj Ahmed
25. Anir Kti
26. Anurag Hussain
27. Rahul Kalita
28. Mostaf Ahmed
29. Sanjib Borbaruah
30. Dhruva Jyoti Talukdar
31. Nishaman Sarma

Essay Competition, Dept of Chemistry 28/5/22

Signature of Teachers

1. Gauran Baisya
2. Mukut Ch Baisya
3. Khalilur Rahman
4. Bishajit Bhaia
5. Mehenala Pathak
6. ~~Dulfer~~ Lalma



(Signature)
 Principal / C & Secretary
 BARBHAG COLLEGE

REPORT ON CLASS SEMINAR
DEPARTMENT OF CHEMISTRY, BARBHAG COLLEGE

DATE: 10/06/22

VENUE: DEPARTMENTAL CLASS ROOM

Today on 10/06/22 a class seminar is held among the Chemistry Honors students of B.Sc 2nd and 4th Semester. Total twenty six (26) no students were participated in the Seminar. The topic of the seminar for 4th Semester students was coordination compounds and the same for the 2nd Semester students was Hybridization. From the 4th Semester Honors Class, Sumita Rajbongshi, Himangshu Patowary & Manashjyoti Baishya presented their Seminar topic. Again from the 2nd Semester Honors Class Naba deep Lahkar, Chandrasmita Sarma and Chandrama Talukdar presented their Seminar Topic. All faculty members of the Chemistry Department namely Mr. Gautam Baishya, Mukut Ch. Baishya , Md. Khalilur Rahman and Mr. Biswajit Bhuyan were present in the Seminar as resource person . After the presentation, an interactive session was held among the participants & resource person. All the students presented their topic elaborately. At the end of Seminar the resource persons suggested them to follow the rules & regulation of Seminar in future course of time.

Photographs and signature sheet are attached herewith.




Principal I/c & Secretary
Barbhag College



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class Seminar 10/06/2022

Dept. of Chemistry, Barbhag college for B.Sc.(H) 2nd & 4th Sem

1. Himangshu Patowary.	B.Sc 4th Sem (H)
2. Nabadeep Kalita.	B.Sc 4th Sem (H)
3. Manash Jyoti Barishya.	B.Sc 4th Sem (H)
4. Bibek Masumdar	B.Sc 4th Sem (H)
5. Rupjyoti Basuman	B.Sc 4th Sem (H)
6. Sumita Borborghi	B.Sc. 4th Sem (H)
7. Chandrama Talukdar	B.Sc. 2nd Sem (H)
8. Chandrasmita Sarma	B.Sc. 2nd Sem (H)
9. Ranjee Rahman	B.Sc 2nd Sem (H)
10. Nabadeep Lohkar.	B.Sc 2nd sem (H)
11. Bhaskar Jyoti Das.	B.Sc. 2nd Sem (H)
12. Naba Jyoti Deka	B.Sc. 2nd Sem (H)
13. Jukin Kumar Sarma	B.Sc. 2nd sem (H)
14. Nandana Begum	B.Sc 2nd sem (H)
15. Patishmita Begum	B.Sc 2nd Sem (H)
16. Krishna Lohkar	B.Sc 2nd Sem (H)
17. Minakshi Barishya	B.Sc 2nd sem (H)
18. Dhritiman Sarma	B.Sc 2nd sem (H)
19. Pranjal Kler	B.Sc 2nd Sem (H)
20. Tanvi Zaman	B.Sc 2nd Sem (H)
21. Madhusmita Das	B.Sc 4th Sem "
22. Anurag Kashyap	B.Sc 2nd Sem (H)
23. Anind Hazar.	B.Sc 4th Sem (H)
24. Nishinjan Balukdar	B.Sc 2nd Sem
25. Nayan Jyoti Deka.	B.Sc 4th Sem
26. Bishal Kalita	B.Sc 2nd Sem

Signature of teachers attended

1. Goutam Barishya.
2. Mukut Ch Barbhag
3. Bishanjit Bha.
4. Pranamita Pathak
5. Lutfu Rahman.



Rinshi

Principal I/c & Secretary
 Barbhag College

EXTEMPORE SPEECH COMPETITION

Organized by

Department of Physics, Barbhag College

Date: 13- 06-22

An extempore speech competition was organized by the department of Physics, Barbhag College among the students of the college on 13-06-2022. The competition was held in the conference hall. The objective of the competition was to make the students confident and better orators. The competition was inaugurated by Dr. Karuna Baruah, Academic In-charge of the college. In his inaugural speech Dr. Baruah said that extempore speech is an excellent way to assist the students not only in spontaneous thinking but also in asserting their creative ideas with precision. He ended his lecture conveying best wishes to the participants. The competition was started at 12:30 pm. The judges were Mr. Dweepen Kr. Das, HOD, Botany and Mr. Lakshi Nath Coudhury, Assistant professor, department of physics. Number of students participated in the competition was 15. The winners were-

Name	Class	Position
Riya Devi	B.A. 1 st Semester	First
Anindita Baruah	B.A. 3 rd Semester	Second
Satirtha Thakuria	B.Sc 3 rd Semester	Third

Prizes were distributed at the end of the programme.





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Ranjana

Principal i/c & Secretary
Barbhag College



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SIGNATURE:

Extempore Speech Competition
 Department of Physics
 Date: 16/06/2022 Venue: Conference Hall

Judges:

- Dwijen Kumar Das, Assistant Professor
 Department of Botany, Barbhag College.
D. 16/6/22
- Lakshi Nath Choudhury,
 Assistant Professor, Department
 of Physics, Barbhag College.
D. 16/6/22

Guest: Dr. Koruna Boruah
 Academic In-charge, Barbhag College.
 Koruna Boruah 16/06/22

Participants:

- Raktim Talukdar BSc. 2nd Sem.
- Nabadeep Lohkora BSc. 2nd Sem.
- Jiemani Deka BSc. 2nd Sem.
- Riya Devi BSc. 2nd Sem.
- Pratham Baruah BSc. 2nd Sem. (Nalbari College)
- Sajid Ahmed BSc. 4th Sem.
- Bhargab Patowary HS. 1st year
- Anindita Baruah 2nd Sem.
- Pranabkrisna Bhuyan 2nd Sem.
- Anurag Kashap HS. Final Candidate
- Satvitha Thakuria BSc. 2nd Sem.
- Jahnali Kalita BSc. 2nd Sem.
- Nilottpal Deka BSc. 2nd Sem.
- Nirban Das BSc. 4th Sem.
- Nakibur Saman BSc. 6th Sem.

16. Tyatormoy Talukdar BSc. 2nd Sem.

Signature:

- Sumi Begum
- Soni Begum
- Nishita Kalita
- Nijuma Begum
- Sri Sangita Kalita
- Mousumi Begum
- Niya Das
- Satobdi Boinya
- Rina Devi
- Anindita Baruah
- Pranabkrisna Bhuyan
- Anurag Kashap
- Pranabkrisna Kalita
- Satvitha Thakuria
- Nabadeep Deka
- Pranabkrisna Bhuyan
- Bishal Kalita
- Jubin Kumar Sarma
- Bhargab Kalita
- Abdul Latif
- Bishal Kalita
- Pouloma Baruah
- Korakjyoti Bas.
- Kausthik Deka

Rinichi
 Principal /c & Secretary
 Barbhag College

Mahananda Pathak
 Head of the Deptt-of Physics
 Barbhag College, Nalbari (Assam)